

La Cumbre Mutual Water Company
695 Via Tranquila Santa Barbara 967-2376
2024 CONSUMER CONFIDENCE REPORT DATA

SOURCE WATER

Este informe contiene información muy importante sobre su agua beber. Tradúzcalo ó hable con alguien que lo entienda bien.

Parameter	Units	State MCL	PHG (MCLG)	Range Average	GROUND WATER	SURFACE WATER	SPILL WATER	Major Sources in Drinking Water
PRIMARY STAND	ARDSI	Mandatory He	alth-Related S					,
CLARITY (a)								
Combined Filter	NTU	TT=1 NTU		ghest Single	NIA	0.09	0.11	Soil runoff
Effluent Turbidity		TT=95% of samp		leasurement	NA NA	100% Sample	0.11 s <= 0.3 NTU	
MICROBIOLOGICAL (b)				,			
Total Coliform		5% of						
Bacteria	Samples		(0)	Reporting	O Desitives	0.700/	O Desitives	Naturally present in the environment
(Distribution System) Fecal Coliform and		samples (b)		Value Range	0 Positives 0 Positives	0.73% 0 Positives	0 Positives 0 Positives	
E. coli	Samples	(b)	(0)	Average	0 Positives	0 Positives		Human and animal fecal waste
(Distribution System)				Highest	0 Positives	0 Positives	0 Positives	
Disinfectant Byproduc	ts Disinfe	ectant Residuals	- Disinfection By					
Total Trihalomethanes	nnh	80	NA	Range	26.60 - 54.70 (d)	3.2 - 47 34	21 - 47 44	By-product of drinking water disinfection
<i>(Distribution System)(c)</i> Haloacetic Acids (c)	ppb	80	INA	Average Range	41.20 (d) 7.4 - 26.7 (d)	0.44 - 21	11 - 24	By-product of drinking water
(Distribution System)	ppb	60	NA	Average	13.1 (d)	16.8	20	disinfection
Disinfectant - Free		MRDL as Cl2	MRDLG as CI2	Range	0.20 - 1.80 (d)	ND - 1.81	0.45 - 1.93	Measurement of the disinfectant
Chlorine Residual Control of DBP pre-	ppm	4.0	4.0	Average Range	0.82 (d) 0.57 - 1.70 (d)	0.82 1.29 - 2.22	1.23 1.6 - 2.2	used in the production of drinking water TOC has no health effects. However, it provides
cursors - TOC	ppm	TT	NA	Average	1.03 (d)	1.69	1.9	a medium for the formation of disinfection by-
								products. Various natural and manmade sources
NORGANIC CHEMICA	LS			Danna	ND	NIA	ND 0.11	Decidus from water treatment were
Aluminum	ppb	1000	600	Range Average	ND NA	NA ND	ND - 0.11 0.050	Residue from water treatment process; Erosion of natural deposits
				Range	ND	NA	NA	Erosion of natural deposits
Arsenic	ppb	10	0.004	Average	NA	NA	NA	•
Barium	nnm	1	2	Range Average	ND NA	NA NA	NA NA	Erosion of natural deposits
Januili	ppm	'	۷	Range	0.41	0.395 - 0.49	NA NA	Erosion of natural deposits;
Fluoride	ppm	2	1	Average	0.41	0.44	0.42	water additive for tooth health
				Range	ND - 7.50	0.17 - 0.79	NA	Runoff & leaching from fertilizer
Nitrate (as NO3)	ppm	45	45	Average Range	3.59 ND	0.38 NA	ND NA	use; sewage; natural erosion Discharge from electroplating factories, leather
Chromium	ppb	50	10	Average	ND NA	NA NA	NA NA	tanneries, wood preservation, chemical synthesis,
5111-511114111	PP~			, wo.ago				refractory production, and textile manufacturing
	1			<u> </u>				facilities; erosion of natural deposits
_EAD & COPPER RULI	ES - Mani	tored at the cust	omers tan Numb	er of eitee o	Action I	evel = 0		
LAD & COFFER ROLL	_3 - WOIII	tored at the cust	oniers tap. Numi	Value	Acceded Action L	-6461 - 0		Internal corrosion of household plumbing
Copper	ppm	1.3	0.3	90th %	0.55	NA	NA	systems; erosion of natural deposits;
								leaching from wood preservatives.
and		15	0.0	Value	ND	NA	NA	Internal corrosion of household plumbing
Lead	ppb	15	0.2	90th %	ND	INA	INA	systems; discharges from industrial manufactures; erosion of natural deposits.
								manadado, orosion or natural doposito.
RADIOCHEMISTRY - R	adioactive	Contaminants						
Gross Alpha				Range	ND - 3.40	NA	NA	Erosion of natural deposits
	pCi/L	15	MCLG, 0	Average	1.33	NA	ND	
Radium 228				Range	1.36	NA	NA	Erosion of natural deposits
	pCi/L	N/A	N/A	Average	1.36	NA	NA	
SECONDARY STA	ANDARD	SAesthetic	Standards					
				Range	63 - 130	14.8 - 18	13 - 19	Runoff/leaching from natural deposits;
Chloride	ppm	500	NA	Average	96	16	15	seawater influence
				Range	ND	ND - 5	ND - 5	
Color (ACU)	Units	15	NA	Average	ND ND	4 21 - 59	ND NA	Naturally occurring organic materials
Copper	ppb	1000	NA	Range	ND			Corrosion of plumbing systems; erosion of natural
оорро.	PP~	1000		Average	NA I			deposits: leaching from wood preservatives
ron	ppb			Average Range	NA ND - 390	40 ND	NA NA	deposits; leaching from wood preservatives Leaching from natural deposits;
		300	NA	Range Average	ND - 390 9.67	40 ND ND	NA NA NA	Leaching from natural deposits; industrial wastes
Mongonoss				Range Average Range	ND - 390 9.67 ND - 32	40 ND ND NA	NA NA NA NA	Leaching from natural deposits;
Manganese	ppb	300 50	NA NA	Range Average Range Average	ND - 390 9.67 ND - 32 0.89	40 ND ND NA ND	NA NA NA NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits
-				Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4	40 ND ND NA	NA NA NA NA	Leaching from natural deposits; industrial wastes
Odor Threshold Specific	ppb Units µmho/	50 3	NA NA	Range Average Range Average Range Average Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400	40 ND ND NA ND 8 8 8 838 - 1079	NA NA NA NA NA 1 - 2 1 760 - 1330	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions
Odor Threshold Specific	ppb Units	50	NA	Range Average Range Average Range Average Average Range Average Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300	40 ND ND NA ND 8 8 8 88-1079	NA NA NA NA 1 - 2 1 760 - 1330 920	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence.
Odor Threshold Specific Conductance	ppb Units µmho/ cm	50 3 1600	NA NA NA	Range Average Range Average Range Average Average Range Average Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300	40 ND ND NA ND 8 8 8 838 - 1079 956 232 - 360	NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate	ppb Units µmho/	50 3 1600 500	NA NA NA	Range Average Range Average Range Average Average Range Average Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 300 810 - 840	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760	NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate Total Dissolved	ppb Units µmho/ cm	50 3 1600	NA NA NA	Range Average Range Average Range Average Range Average Range Average Range Average Average Average Range Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 300 810 - 840 825	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659	NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids	ppb Units µmho/ cm ppm	50 3 1600 500	NA NA NA NA	Range Average Range Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 300 810 - 840 825 0.80	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2	NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids	ppb Units µmho/ cm ppm	50 3 1600 500	NA NA NA	Range Average Range Average Range Average Range Average Average Range Average Average Range Average Average Average Average Average Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 300 810 - 840 825	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659	NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly)	ppb Units µmho/ cm ppm	50 3 1600 500	NA NA NA NA	Range Average Range Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 300 810 - 840 825 0.80 0.80	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2	NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly)	ppb Units µmho/ cm ppm ppm NTU ppm	50 3 1600 500 1000 5	NA NA NA NA NA NA NA	Range Average Range Average Range Average Range Average Average Average Average Range Average Range Average Range Average Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 300 810 - 840 825 0.80 ND - 0.75	40 ND ND NA ND 8 838-1079 956 232-360 302 568-760 659 0.15-4.2 1.2	NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11 NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water, seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Turbidity (Monthly) Tinc Additional Parame	ppb Units µmho/ cm ppm ppm NTU ppm	50 3 1600 500 1000 5	NA NA NA NA NA NA NA	Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 0.80 ND - 0.75 0.38	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA	NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11 0.04 NA NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as	ppb Units µmho/ cm ppm ppm NTU ppm	50 3 1600 500 1000 5 5.0	NA NA NA NA NA NA NA NA	Range Average Range Range Average Range Average Range Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38	40 ND ND NA ND 8 8 838-1079 956 232-360 302 568-760 659 0.15-4.2 1.2 NA NA	NA NA NA NA NA 1-2 1 760-1330 920 230-290 260 NA 530 0.02-0.11 0.04 NA NA NA NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as	ppb Units µmho/ cm ppm ppm NTU ppm	50 3 1600 500 1000 5	NA NA NA NA NA NA NA	Range Average Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 300 810 - 840 825 0.80 0.80 ND - 0.75 0.38	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA	NA NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11 0.04 NA NA NA 150 - 190 170	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence
Odor Threshold Specific Conductance Sulfate Fotal Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents	ppb Units µmho/ cm ppm ppm NTU ppm	50 3 1600 500 1000 5 5.0	NA NA NA NA NA NA NA NA	Range Average Range Range Average Range Average Range Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38	40 ND ND NA ND 8 8 838-1079 956 232-360 302 568-760 659 0.15-4.2 1.2 NA NA	NA NA NA NA NA 1-2 1 760-1330 920 230-290 260 NA 530 0.02-0.11 0.04 NA NA NA NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate Fotal Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as	ppb Units µmho/ cm ppm ppm NTU ppm sters (Ur	50 3 1600 500 1000 5 5.0 nregulated):	NA	Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510	40 ND ND NA ND 8 8 838 - 1079 932 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 104 368 - 496	NA NA NA NA NA NA 1-2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11 0.04 NA NA NA NA 150 - 190 170 76 - 93 85 300 - 380	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as	ppb Units µmho/ cm ppm ppm NTU ppm	50 3 1600 500 1000 5 5.0 nregulated):	NA NA NA NA NA NA NA	Range Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 104 368 - 496 426	NA NA NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11 0.04 NA NA NA 150 - 190 170 76 - 93 85 300 - 380	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits; seawater influence Leaching from natural deposits
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Turbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as CaCO3	ppb Units µmho/ cm ppm ppm NTU ppm sters (Ur ppm	50 3 1600 500 1000 5 5.0 NA NA	NA	Range Average Average Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495 36 - 38	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 104 368 - 496 426 28 - 55	NA NA NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 0.02 - 0.11 0.04 NA NA NA NA 150 - 190 76 - 93 85 300 - 380 350 27 - 38	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits; seawater influence Leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as CaCO3	ppb Units µmho/ cm ppm ppm NTU ppm sters (Ur ppm ppm	50 3 1600 500 1000 5 5.0 nregulated):	NA	Range Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495 36 - 38	40 ND ND NA ND 8 8 838 - 1079 932 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 104 368 - 496 426 28 - 55 44	NA NA NA NA NA NA NA NA 1-2 1 760 - 1330 9230 - 290 260 NA 530 0.02 - 0.11 0.04 NA NA NA NA NA SA NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence
Odor Threshold Specific Conductance Sulfate Fotal Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as CaCO3	ppb Units µmho/ cm ppm ppm NTU ppm sters (Ur ppm	50 3 1600 500 1000 5 5.0 NA NA	NA	Range Average Average Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495 36 - 38 37 7.1 - 7.26	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 104 368 - 496 426 28 - 55 44 7.32 - 7.83 7.56	NA NA NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 0.02 - 0.11 0.04 NA NA NA NA 150 - 190 76 - 93 85 300 - 380 350 27 - 38	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as CaCO3 Magnesium DH	ppb Units µmho/ cm ppm ppm NTU ppm sters (Ur ppm ppm ppm ppm ppm ppm ppm ppm ppm pp	50 3 1600 500 1000 5 5.0 NA NA NA NA	NA N	Range Average Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495 36 - 38 37 7.1 - 7.6 7.26 2.7 - 3.4	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 104 368 - 496 426 28 - 55 44 7.32 - 7.83 1.8 - 56 1.8 - 3.0	NA NA NA NA NA NA NA NA 1-2 1 760-1330 920 260 NA 530 0.02-0.11 0.04 NA NA NA NA NA S 150-190 170 76-93 85 300-380 350 27-38 34 6.85-8.43 7.86 NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Furbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as CaCO3 Magnesium DH	ppb Units µmho/ cm ppm ppm NTU ppm Pters (Ur ppm ppm ppm ppm ppm ppm ppm ppm ppm	50 3 1600 500 1000 5 5.0 nregulated): NA NA	NA	Range Average	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495 36 - 38 37 7.1 - 7.6 7.26 2.7 - 3.4 3.05	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 1.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 194.5 - 115 426 28 - 55 44 7.32 - 7.83 7.56 1.8 - 3.0 2.4	NA NA NA NA NA NA NA NA NA 1 - 2 1 760 - 1330 920 260 NA 530 0.02 - 0.11 0.04 NA	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence
Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Turbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as CaCO3 Magnesium pH	ppb Units µmho/ cm ppm ppm NTU ppm sters (UI ppm ppm ppm ppm ppm ppm ppm ppm ppm pp	50 3 1600 500 1000 5 5.0 nregulated): NA NA NA NA NA NA	NA N	Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495 36 - 38 37 7.1 - 7.6 7.26 2.7 - 3.4 3.05 79 - 85	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 104 368 - 496 426 28 - 55 44 7.32 - 7.83 7.56 1.8 - 3.0 2.4 40 - 57	NA NA NA NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11 0.04 NA NA NA 150 - 190 170 76 - 93 85 300 - 380 350 27 - 38 34 6.85 - 8.43 7.86 NA NA 2.2 43 - 54	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;
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Manganese Odor Threshold Specific Conductance Sulfate Total Dissolved Solids Turbidity (Monthly) Zinc Additional Parame Alkalinity (Total) as CaCO3 equivalents Calcium as Ca Hardness (Total) as CaCO3 Magnesium pH Potassium Sodium Constituents of Co	ppb Units µmho/ cm ppm ppm NTU ppm ppm ppm ppm ppm ppm ppm ppm ppm pp	50 3 1600 500 1000 5 5.0 nregulated): NA NA NA NA NA NA	NA N	Range Average Range	ND - 390 9.67 ND - 32 0.89 1 - 4 3 1200 - 1400 1300 300 810 - 840 825 0.80 ND - 0.75 0.38 270 - 340 301.11 130 - 160 146.25 480 - 510 495 36 - 38 37 7.1 - 7.6 7.26 2.7 - 3.4 3.05 79 - 85	40 ND ND NA ND 8 8 838 - 1079 956 232 - 360 302 568 - 760 659 0.15 - 4.2 1.2 NA NA NA 164 - 255 191 94.5 - 115 104 368 - 496 426 28 - 55 44 7.32 - 7.83 7.56 1.8 - 3.0 2.4 40 - 57	NA NA NA NA NA NA NA NA 1 - 2 1 760 - 1330 920 230 - 290 260 NA 530 0.02 - 0.11 0.04 NA NA NA 150 - 190 170 76 - 93 85 300 - 380 350 27 - 38 34 6.85 - 8.43 7.86 NA NA 2.2 43 - 54	Leaching from natural deposits; industrial wastes Leaching from natural deposits Naturally occurring organic materials Substances that form ions when in water; seawater influence. Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; seawater influence Soil runoff Naturally occurring in trace amounts, but can be detected in soft, acidic water systems Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;

Water System Name: La Cumbre Mutual Water Company

In 2024, 44% of our water was from our wells, 20% Lake Cachuma Spill Water and the remaining 36% was State Water. Goleta Water District (GWD) treats our spill water. The City of Santa Barbara treats our State Water. The State Water quality portion of this report comes from the City of Santa Barbara and Spill Water from GWD. Sections of our service area along State Street receive water that was treated entirely by the City of Santa Barbara or GWD.

and place of regularly scheduled board meetings for public participation: Once a month at 695 Via Tranquila, pla se call for exact date and times 805 967-2376.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include

- Contaminants trait may be present in source water include:

 Microbial contaminants, such as viruses and bacteria. that may come from sewade treatment plants, septic systems, agricultural livestock operations, and wildlife.

 Inorganic contaminants, such as salts and metals, that can be naturally-occurring or result from urban storm water runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.

 Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff, and residential uses.

 Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban storm water runoff, and septic systems.

 Radioactive contaminants, which can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, USEPA and the State Water Resources Control Board (SWRCB) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. SWRCB regulations also establish limits for contaminants in bottled water that must provide the same protection for public health.

This report lists all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The SWRCB requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants are not expected to vary significantly from year to year. Some of the data, though representative of the water quality, are more than one year old.

Additional General Information On Drinking Water:

All drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline at 1-800-426-4791 or www.epa.gov/safewater/.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice from their health care providers about drinking water. USEPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline 1-800-426-4791 or www.epa.gov/safewater/.

ABBREVIATIONS AND NOTES

Type of water sources: Five ground water wells, Spill Water treated by GWD and State Water from Lake Cachuma through SB City Cater Treatment Plant.

Name of Sources: Well #16, Well #17, Well #18, Well #19, Well #21 & Well #22 and seven metered connections to Santa Barbara City Water. Note: Depending on where you live, our water is a mixture of groundwater, spill water and surface water.

Water Quality Report: Listed are substances detected in the drinking water. Not listed are more than 135 regulated and unregulated substances that were belo the laboratory detection level.

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materiand components associated with service lines and home plumbing. La Cumbre Water Co. is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

Definitions:

Public Health Goal (PHG): The level of a contaminant in drinking water

Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the Environmental Protection Agency.

Maximum Contaminate Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency

MCLGs are set by the U.S. Environmental Protection Agency

Maximum Contaminate Level (MCLs): The highest level of a contaminant
that is allowed in drinking water. Primary MCLs are set as close to the
PHGs (or MCLGs) as is economically and technologically feasible.
Secondary MCLs are set to protect the odor, taste, and appearance of
drinking water.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a
disinfectant (chlorine) added for water treatment below which there is no

disinfectant (chlorine) added for water treatment below which there is no known or expected risk to health. MRDLGs are set by the U. S.

Environmental Protection Agency.

Maximum Residual Disinfectant Level (MRDL): The level of a disinfectant (chlorine) added for water treatment that may not be exceeded at the consumer's tap.

Regulatory Action Level (AL): The concentration of a contaminant which if exceeded, triggers a treatment or other requirement which a water system must follow.

Treatment Technique (TT): A required process intended to reduce

the level of a contaminant in drinking water.

Primary Drinking Water Standards (PDWS): MCLs and MRDLs for

contaminants that affect health along with their monitoring and reporting requirements and water treatment requirements.

requirements and water treatment requirements.

Secondary Drinking Water Standards (SDWS): MCL's for contaminants that effect taste, odor or appearance of drinking water. Contaminants with SWDS do not affect the health at MCL levels.

Unregulated Contaminant Monitoring Regulations (UCMR): Data generated

by the new UCMR will be used to evaluate and prioritize contaminants on the Drinking Water Contaminant Candidate List, a list of contaminants EPA is considering for possible new drinking water standards. Also known as "State Regulated Contaminants with No MCLs".

Footnotes:

- (a) Turbidity (NTU) is a measure of the cloudiness of the water and it is a good indicator of the effectiveness of our filtration system.

 Monthly turbidity values for ground water are listed in the Secondary Standards section
- (b) Total coliform MCLs: The State MCL for coliforms is no more than 1 (b) Total coliform MCLs: The State MCL for coliforms is no more than 1 per month for water systems which collect less than 40 samples per month (La Cumbre Water). Systems which collect over 40 routine samples may not have more than 5% positive per month.
 (c) Compliance based on the quarterly annual average distribution system samples.
 (d) Although reported under ground water these readings were taken from the distribution system and are a combination of ground and surface water.

Abbreviations

<" = Less Than

"<" = Less Than

AL = Regulatory Action Level

ACU = Apparent Color Units

DBP = Disinfection Byproducts

DNQ = Detected, not Quantifiable

MCL = Maximum Contaminant Level

MCLG = Maximum Contaminant Level

MRDL = Maximum Residual Disinfectant Level

MRDLG = Maximum Residual Disinfectant Goal

MRDL = Maximum Residual Disinfectant Goal

NA = not applicable

NC = Not Collected

ND = None Detected

NTU = Nephelometric Turbidity Units

NTU = Nephelometric Turbidity Units

NTU = Nephelometric Turbidity Units
pCi/L = PicoCuries per liter
PHG = Public Health Goal
ppb = parts per billion, or micrograms per liter (μg/L)
ppm = parts per million, or milligrams per liter (mg/L)
TOC = Total Organic Carbon
TT = Treatment Technique
μmho/cm = micromhos per centimeter
(unit of specific conductance of water)